Synthesis and structure elucidation of hydrazones derived from \(N-(2,4\text{-dimethylphenyl})-3\text{-oxobutanamide}\)

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Abstract
Diazonium salts derived from amines 1 (sulfanilic acid, 4-nitroaniline, 4-aminosalicylic acid, sulfanilamide, sulfamethoxazole, sulfathiazole, sulfapyridine, sulfamerazine) were coupled with \(N-(2,4\text{-dimethylphenyl})-3\text{-oxobutanamide}\) (2) resulting in the formation of hydrazones 3a–h.

Keywords: Azo coupling, hydrazones

Introduction

Coupling products of diazonium salts with aliphatic active hydrogen compounds are widely used as intermediates for the synthesis of a large number of heterocyclic compounds. Pyrazoles,\(^1,2\) isoxazolone,\(^3\) 2-pyrazoline-5-one\(^4,5\) can be obtained by cyclization of coupling products with substituted hydrazine or hydroxylamine, respectively (Scheme 1). Both hydrazones\(^5,6-9\) and their cyclization compounds\(^2-4\) possess important biological and pharmacological properties.

Scheme 1. General synthetic routes for the formation of azo compounds and hydrazones.
Azo dyes are among the most important synthetic coloring agents but are regarded as potential carcinogens\(^\text{10}\) due to their metabolism; the reduction of the azo group affects their toxicity, mutagenicity, and carcinogenicity.\(^\text{10-13}\) Hydrazone dyes are considered non-genotoxic and non-mutagenic; e.g., 2-[(2-methoxy-5-nitrophenyl)hydrazono]-N-(2-methoxyphenyl)-3-oxobutanamide (PY74) is a hydrazone pigment used in yellow tattoo inks. The metabolism of PY74\(^\text{14}\) and compounds containing azo group\(^\text{15,16}\) has been investigated using rat liver and human liver microsomes.

![PY74](image)

The hydrazone product obtained by azo coupling of the diazonium salt of sulfapyridine with \(N\)-(2-methylphenyl)-3-oxobutanamide has been found to be an HIV integrase inhibitor.\(^\text{17}\)

![HIV integrase inhibitor](image)

Earlier reports indicate that diazonium salts of certain aromatic amines such as sulfaguanidine\(^\text{18}\) and sulfanilamide\(^\text{19}\) have been coupled with compounds possessing active hydrogen. Furthermore, sulfanilamide derivatives have been reported to possess antibacterial activity.\(^\text{20}\) The present study reports on the synthesis of new coupling products 3a–h, which were obtained from diazonium salts derived from amines 1 (sulfanilic acid, 4–nitroaniline, 4–aminosalicylic acid, sulfanilamide, sulfamethoxazole, sulfathiazole, sulfapyridine, sulfamerazine) with \(N\)-(2,4-dimethylphenyl)-3-oxobutanamide (2).
Results and Discussion

The diazonium salts derived from anilines 1 (sulfanilic acid, 4–nitroaniline, 4–aminosalicylic acid, sulfanilamide, sulfamethoxazole, sulfathiazole, sulfapyridine, sulfamerazine) were coupled with N-(2,4-dimethylphenyl)-3-oxobutanamide (2) in aqueous ethanol containing sodium acetate resulting in the formation of hydrazones 3a–h (Scheme 2).

The UV spectra of products 3 show three ranges of absorption maxima at 203–207, 235–271, and 376–393 nm, except compounds 3e and 3f which had four absorption maxima. Absorption bands attributed to an azo function between 332–360 nm and above 400 nm are missing. Thus, the observation of bands at 376–393 nm is indicative of the hydrazone form of compounds 3a–h.

Scheme 2. Preparation of [2-[1-(2,4-dimethylphenylamino)-1,3-dioxobutan-2-ylidene]hydrazinyl]benzenes (3a–h).

The amide proton (-NH-C=O) exhibits a singlet at δ 10.98–11.16, the hydrazone proton (-CH=N-NH-) shows a singlet at δ 13.90–14.32; both signal ranges are in agreement with the literature. Furthermore, in the 1H-NMR spectra of compounds 3a–h, signals arising from a >CH-N=N- moiety are expected at 3.00–4.00 ppm, but were not observed. These findings support the hydrazone structure of the products.

The APCI-MS spectra of 3a–h show molecular ion peaks (M+) confirming their molecular weight; common characteristic fragment ions result from cleavage of the amide bond resulting in
2,4-dimethylanilinium ion (m/z 122) and the complementing 2-[(2-arylhydrazono)-3-oxo-butylidyne]oxonium ion (m/z M+1–122).

**Experimental Section**

**General Procedures.** All chemicals and solvents were commercially acquired. Melting points were determined with a Schmelzpunktbestimmer SMP II. The UV spectra were measured with a Shimadzu UV–2100 S. The IR spectra were obtained with a Shimadzu FTIR–8400. 1H-NMR spectra in DMSO-d6 were recorded on a Bruker Avance-DPX-400 spectrometer (400 MHz). APCI-MS was performed using an Agilent 1100 MSD spectrometer at 100 eV (positive polarity). All new compounds were analyzed for C, H, N (Leco CHNS 932).

1H NMR, APCI-MS, and elemental analyses were provided by the Scientific and Technical Research Council of Turkey, (TÜBITAK).

[2-[1-(2,4-Dimethylphenylamino)-1,3-dioxobutan-2-ylidene]hydrazinyl]benzenes (3a–h).

**General procedure**

To the cooled (0–5 °C) solution of amine 1 (0.01 mol) in ethanol (50 mL) and hydrochloric acid (4%; 40 mL) was added an ice-cold solution of sodium nitrite (7%; 10 mL). The reaction mixture was then poured into a solution of N-(2,4-dimethylphenyl)-3-oxo–butanamide (2; 2.05 g, 0.01 mol) and sodium acetate (60 g, 0.73 mol) in ethanol (50%; 50 mL) under vigorous stirring. The precipitated solid was collected, washed with water, air-dried at room temperature, and washed with ethanol to give 3.

1H NMR spectra: H’ refers to the X,Y-substituted benzene ring Ar (cf. Scheme 2).

**4-[2-[1-(2,4-Dimethylphenylamino)-1,3-dioxobutan-2-ylidene]hydrazinyl]benzenesulfonylic acid (3a).** Yellow needles (3.46 g, 89%); mp 357 °C (EtOH; decomp.). UV $\lambda_{max}$ (EtOH): nm (log $\varepsilon$) 376 (4.53), 250 (4.26), 204 (4.45). IR (KBr): $\nu_{max}$ (cm$^{-1}$): 3592, 3511 (OH), 3170–3130 (NH), 1680 (ketone C=O), 1600 (amide C=O), 1315–1135 (S=O). 1H NMR (DMSO-d6): $\delta$ 2.26 (3H, s, CH$_3$), 2.29 (3H, s, CH$_3$), 2.55 (3H, s, COCH$_3$), 7.04 (1H, d, $^3J_{6,5} = 8.2$ Hz, H$_6$), 7.11 (1H, s, H3), 7.45–7.70 (4H, m, H2', H3', H5', H6'), 7.96 (1H, d, $^3J_{5,6} = 8.2$ Hz, H5), 11.16 (1H, s, CONH), 14.32 (1H, s, NNH). APCI-MS: m/z (%) 390 (100) [M+1]$^+$, 371 (3.7), 269 (24), 152 (4.4), 122 (37). Anal. calcd. for C$_{18}$H$_{19}$N$_3$O$_5$S·2H$_2$O: C, 50.81; H, 5.45; N, 9.88; S, 7.54. Found: C, 50.25; H, 5.24; N, 10.03; S, 7.23.

**N-(2,4-Dimethylphenyl)-2-[2-(4-nitrophenyl)hydrazono]-3-oxobutanamide (3b).** Yellow needles (2.76 g, 78%); mp 222 °C (EtOH). UV $\lambda_{max}$ (EtOH): nm (log $\varepsilon$) 392 (4.40), 235 (4.06), 203 (4.34). IR (KBr): $\nu_{max}$ (cm$^{-1}$): 3220, 3130 (NH), 1663 (ketone C=O), 1595 (amide C=O). 1H NMR (DMSO-d6): $\delta$ 2.27 (3H, s, CH$_3$), 2.29 (3H, s, CH$_3$), 2.57 (3H, s, COCH$_3$), 7.05 (1H, d, $^3J_{6,5} = 8.8$ Hz, H6), 7.11 (1H, s, H3), 7.71–8.30 (5H, m, H5, H2', H3', H5', H6'), 11.00 (1H, s, CONH), 13.90 (1H, s, NNH). APCI-MS: m/z (%) 355 (100) [M+1]$^+$ 234 (31.5), 139 (5.0), 122

4-[2-[1-(2,4-Dimethylphenylamino)-1,3-dioxobutan-2-yldene]hydrazinyl]-2-hydroxybenzolic acid (3c). Yellow needles (2.92 g, 79%); mp 245 °C (EtOH). UV λₘₐₓ. (EtOH): nm (log ε) 386 (4.55), 271 (4.24), 206 (4.60). IR (KBr): νₘₐₓ. (cm⁻¹): 3220, 3130 (NH), 1680 (carboxylic acid C=O), 1660 (ketone C=O), 1638 (amide C=O). ¹H NMR (DMSO-d₆): δ 2.26 (3H, s, CH₃), 2.28 (3H, s, CH₃), 2.54 (3H, s, COCH₃), 7.00 (1H, s, H₂'), 7.01 (1H, d, ³J₆,₅ = 8.0 Hz, H₆), 7.05 (1H, d, ³J₆',₅' = 10.0 Hz, H₆'), 7.09 (1H, s, H₃), 7.79 (1H, d, ³J₅,₆' = 9.2 Hz, H₅'), 7.93 (1H, d, ³J₅,₆ = 8.2 Hz, H₅), 11.04 (1H, s, CONH), 13.90 (1H, s, NNH). APCI-MS: m/z (%): 370 (100) [M+1]⁺, 355 (16.6), 249 (53.9), 122 (23.6). Anal. calcd. for C₁₉H₁₉N₃O₅.H₂O: C, 58.91; H, 5.46; N, 10.85. Found: C, 58.73; H, 4.73; N, 11.12.

N-(2,4-Dimethylphenyl)-3-oxo-2-[2-(4-sulfamoylphenyl)hydrazono]butanamide (3d). Yellow needles (3.06 g, 79%); m p 260 °C (EtOH). UV λₘₐₓ. (EtOH): nm (log ε) 376 (4.17), 236 (3.92), 203 (4.33). IR (KBr): νₘₐₓ. (cm⁻¹): 3317, 3234, 3170 (NH), 1663 (ketone C=O), 1596 (amide C=O), 1151 (S=O). ¹H NMR (DMSO-d₆): δ 2.27 (3H, s, CH₃), 2.29 (3H, s, CH₃), 2.57 (3H, s, COCH₃), 7.05 (1H, d, ³J₆,₅ = 8.2 Hz, H₆), 7.10 (1H, s, H₃), 7.34 (1H, s, SO₂NH), 7.66–7.90 (4H, m, H₂', H₃', H₅', H₆'), 7.94 (1H, d, ³J₅,₆ = 8.2 Hz, H₅), 11.05 (1H, s, CONH), 14.10 (1H, s, NNH). APCI-MS: m/z (%): 389 (28.2) [M+1]⁺, 372 (7.3), 283 (43.9), 268 (91.6), 122 (100). Anal. calcd. for C₁₈H₂₀N₄O₄S: C, 55.66; H, 5.19; N, 14.42; S, 8.25. Found: C, 55.83; H, 5.26; N, 14.19; S, 7.84.

N-(2,4-Dimethylphenyl)-2-[2-[4-[N-(5-methylisoxazol-3-yl)sulfamoyl]phenyl]hydrazono]-3-oxobutanamide (3e). Yellow needles (4.03 g, 86%); mp 215 °C (EtOH). UV λₘₐₓ. (EtOH): nm (log ε) 376 (4.44), 249 (4.15), 237 (4.16), 204 (8.98). IR (KBr): νₘₐₓ. (cm⁻¹): 3220, 3130 (NH), 1667 (ketone C=O), 1614 (amide C=O), 1151 (S=O). ¹H NMR (DMSO-d₆): δ 2.26 (3H, s, CH₃), 2.28 (3H, s, CH₃), 2.30 (3H, s, CH₃ at isoxazole), 2.55 (3H, s, COCH₃), 6.15 (1H, s, CH isoxazole), 7.04 (1H, d, ³J₆,₅ = 8.3 Hz, H₆), 7.10 (1H, s, H₃), 7.67–7.92 (4H, m, H₂', H₃', H₅', H₆'), 7.94 (1H, d, ³J₅,₆ = 8.2 Hz, H₅), 10.98 (1H, s, CONH), 11.40 (1H, s, SO₂NH), 14.02 (1H, s, NNH). APCI-MS: m/z (%): 470 (100) [M+1]⁺, 372 (19.6), 349 (48.7), 254 (26.5), 122 (22.0). Anal. calcd. for C₂₂H₂₃N₅O₅S·¹/₂H₂O: C, 56.28; H, 4.94; N, 14.92; S, 6.63. Found: C, 56.19; H, 4.52; N, 14.98; S, 6.63.

N-(2,4-Dimethylphenyl)-3-oxo-2-[2-[4-(N-thiazol-2-ylsulfamoyl)phenyl]hydrazono]butanamide (3f). Yellow needles (3.34 g, 71%); mp 280 °C (EtOH). UV λₘₐₓ. (EtOH): nm (log ε) 376 (4.41), 271 (4.08), 251 (4.03), 202 (4.39). IR (KBr): νₘₐₓ. (cm⁻¹): 3220, 3130 (NH), 1665 (ketone C=O), 1630 (amide C=O), 1149 (S=O). ¹H-NMR (DMSO-d₆): δ 2.26 (3H, s, CH₃), 2.28 (3H, s, CH₃), 2.55 (3H, s, COCH₃), 6.84 (1H, d, ³J₅,₄: 4.5 Hz, H₅ thiazole), 7.03 (1H, d, ³J₆,₅ = 8.3 Hz, H₆), 7.09 (1H, s, H₃), 7.26 (1H, d, ³J₄,₅ = 4.6 Hz, H₄ thiazole), 7.62–7.88 (4H, m H₂', H₃', H₅', H₆'), 7.93 (1H, d, ³J₅,₆ = 8.2 Hz, H₅), 11.03 (1H, s, CONH), 12.75 (1H, s, SO₂NH), 14.11 (1H, s, NNH). APCI-MS: m/z (%): 472 (100) [M+1]⁺, 372 (9.5), 351 (17.0), 256 (10.9), 122 (7.3). Anal. calcd. for C₁₉H₁₉N₄O₄S·¹/₂H₂O: C, 52.50; H, 4.58; N, 14.58; S, 13.33. Found: C, 52.13; H, 5.14; N, 14.83; S, 13.38.
N-(2,4-Dimethylphenyl)-3-oxo-2-[2-[4-(N-pyridin-2-ylsulfamoyl)phenyl]hydrazono]butanam ide (3g). Yellow needles (3.12 g, 67%); mp 240 °C (EtOH). UV $\lambda_{\text{max}}$ (EtOH): nm (log e): 375 (4.11), 247 (3.93), 203 (4.31). IR (KBr): $\nu_{\text{max}}$ (cm$^{-1}$): 3220, 3130 (NH), 1668 (ketone C=O), 1633 (amide C=O), 1139 (S=O). $^1$H NMR (DMSO-$d_6$): $\delta$ 2.26 (3H, s, CH$_3$), 2.28 (3H, s, CH$_3$), 2.54 (3H, s, COCH$_3$), 6.88 (1H, t, H4 pyridine), 7.03 (1H, d, $^3$J$_{6,5}$ = 8.7 Hz, H6), 7.09 (1H, s, H3), 7.16 (1H, d, $^3$J$_{5,6}$ = 8.3 Hz, H5 pyridine), 7.62-7.93 (5H, m, H3 pyridine, H2', H3', H5', H6'), 7.97 (1H, d, $^3$J$_{5,6}$ = 8.3 Hz, H5), 8.01 (1H, d, $^3$J$_{6,5}$ = 4.4 Hz, H6 pyridine), 11.02 (1H, s, CONH), 11.90 (1H, s, SO$_2$NH), 14.08 (1H, s, NNH). APCI-MS: m/z (%) 466 (100) [M+1]$^+$, 345 (3.6), 318 (1.4), 268 (2.2), 250 (4.2), 122 (4.1). Anal. calcd. for C$_{23}$H$_{23}$N$_5$O$_4$S·H$_2$O: C, 57.13; H, 5.21; N, 14.48; S, 6.63. Found: C, 57.95; H, 4.81; N, 14.81; S, 6.85.

N-(2,4-Dimethylphenyl)-2-[2-[4-[N-(4-methylpyrimidin-2-yl)sulfamoyl]phenyl]hydrazono]-3-oxobutanamide (3h). Yellow needles (3.02 g, 63%); mp 288 °C. UV $\lambda_{\text{max}}$ (EtOH): nm (log e): 376 (4.49), 253 (4.28), 203 (4.53). IR (KBr): $\nu_{\text{max}}$ (cm$^{-1}$): 3250, 3120 (NH), 1668 (ketone C=O), 1615 (amide C=O), 1343, 1175 (S=O). $^1$H NMR (DMSO-$d_6$): $\delta$ 2.26 (3H, s, CH$_3$), 2.28 (3H, s, CH$_3$), 2.33 (3H, s, CH$_3$ at pyrimidine), 2.55 (3H, s, COCH$_3$), 6.90 (1H, d, $^3$J$_{6,5}$ = 5.0 Hz, H6 pyrimidine), 7.00 (1H, d, $^3$J$_{5,6}$ = 8.2 Hz, H6), 7.09 (1H, s, H3), 7.65–8.06 (5H, m H5, H2', H3', H5', H6'), 8.32 (1H, d, $^3$J$_{5,6}$ = 5.0 Hz, H5 pyrimidine), 11.01 (1H, s, CONH), 11.67 (1H, s, SO$_2$NH), 14.07 (1H, s, NNH). APCI-MS: m/z (%) 481 (100) [M+1]$^+$, 467 (4.0), 332 (6.3), 265 (4.1). Anal. calcd. for C$_{23}$H$_{24}$N$_6$O$_4$S: C, 57.49; H, 5.03; N, 17.49; S, 6.67. Found: C, 57.36; H, 5.73; N, 17.67; S, 6.59.

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**References**