6-Hydrazinonicotinic acid hydrazide: a useful precursor for chemo- and regioselective synthesis of new heteroaryl-linked pyridinohydrazones

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Abstract

In order to demonstrate the reactivity differentiation between the two dinucleophilic centers in the 6-hydrazinonicotinic acid hydrazide hydrate (**1**), a series of seven new 6-[(2-(arylidene)hydrazinyl)nicotinohydrazides (**4**) was chemoselectively accessed from the reaction of **1** with seven aryl or heteroaryl aldehydes by a simple procedure in ethanol as solvent. In a subsequent study, cyclization reactions of **4** with some 4-alkoxy-1,1,1-trifluoroalk-3-en-2-ones or triethylorthoacetate were proceeded easily also in ethanol, an eco-friendly solvent, to give regioselectively nine examples of pyrazolinyl-pyridinohydrazones and one 1,3,4-oxadiazole derivative, respectively, as stable and pure compounds in good yields.

Keywords: Hydrazones, hydrazides, pyrazoles, oxadiazoles, pyridines

Introduction

Hydrazones and their derivatives constitute a versatile class of compounds in organic chemistry and they are characterized by the presence of -NH-N=CH- or $\text{-NH-N=CR}^1\text{R}^2\text{-}$ groups in their molecules.^{1,2} These compounds have interesting biological properties, such as anti-inflammatory, analgesic, anticonvulsant, antituberculous, antitumor, anti-HIV and antimicrobial activity.³⁻⁶ Hydrazones are also important compounds for drug design, as possible ligands for metal complexes, organocatalysis, and also for the syntheses of heterocyclic compounds.^{7,8} The ease of preparation, increased hydrolytic stability relative to imines, and tendency toward crystallinity are all desirable characteristics of hydrazones. Due to these positive traits, hydrazones have been under study for a long time, but much of their basic chemistry remains unexplored. On the other hand, the presence of heterocyclic rings in many synthetic hydrazones plays a main role in their pharmacological properties.⁸

Pyrazoles have been studied for over a century as an important class of heterocyclic compounds and still attract considerable attention due to the broad range of biological activities, including analgesic, antimicrobial, antiviral, anti-inflammatory, hypoglycemic, anti-hypertensive and antitumor properties.⁹⁻¹³ Pyridine, which also deserves attention, is not found free in nature, but its derivatives have many chemical applications, mostly in polymers and dyes, as well as in the pharmaceutical area as analgesics, anesthetics, and psychopharmacological drugs.¹⁴

The structure of 6-hydrazinonicotinic hydrazide hydrate (**1**) contains two nucleophilic centers: a hydrazine and a hydrazide moiety, which must have differentiated reactivity towards the electrophiles. Thus, in one of our recent studies we carried out the cyclocondensation reactions of 4-alkoxy-4-alkyl(aryl/heteroaryl)-1,1,1-trifluoroalk-3-en-2-ones with 6-hydrazinonicotinic acid hydrazide hydrate, in a equimolar ratio, and obtained regioselectively a new trifluoromethylated system as 2-(1*H*-pyrazolyl)-5-(1*H*-pyrazolylcarbonyl)pyridines in good yields (Figure 1).¹⁵ However, these reaction conditions do not lead to a chemoselective cyclization at the most reactive center (hydrazine moiety).

Figure 1. Generic structures of 4-alkoxy-4-alkyl(aryl/heteroaryl)-1,1,1-trifluoroalk-3-en-2-ones (**I**) and 2-(1*H*-pyrazolyl)-5-(1*H*-pyrazolylcarbonyl)pyridines (**II**).

In this context, the aim of this paper is to report the reactivity study on 6-hydrazinonicotinic acid hydrazide hydrate and to prove it as a versatile precursor for some new interesting heterocycles. Thus, attempting to demonstrate the reactivity differentiation between the two dinucleophilic centers in the hydrazide hydrate (**1**), we report (*i*) the results of the reactions of **1** with triethylorthoacetate (Scheme 1); (*ii*) the synthesis of a new series of 6-(alkylidenehydrazinyl)nicotinohydrazides (**4**) conveniently accessed from the reaction of **1** with aryl or heteroaryl aldehydes (**3**) (Scheme 2), and (*iii*) a subsequent study, where cyclization reactions of **4** with 4-alkoxy-1,1,1-trifluoroalk-3-en-2-ones (**5**) and triethylorthoacetate are proceeded to give examples of new heteroaryl-pyridinohydrazones (**6**) and one 1,3,4-oxadiazole derivative (**7**) (Scheme 3).

Results and Discussion

Synthesis and structure

Firstly, it is well-known from our previous work¹⁵ that 4-alkoxy-1,1,1-trifluoroalk-3-en-2-ones, a 1,3-dielectrophile block, is unable to differentiate the two nucleophilic centers at the compound

1. Thus, in an attempt to demonstrate the reactivity differentiation between the two dinucleophilic centers in the 6-hydrazinonicotinic acid hydrazide hydrate (**1**), we performed the reaction involving **1** and triethylorthoacetate, which was also used as the reaction solvent, at reflux (145 °C) for 16 h (Scheme 1).¹⁶ Unfortunately, no chemical differentiation was observed for compound **1** toward the cited orthoester. In this case the hydrazine and the hydrazide moieties showed similar reactivity and conducted only to the not yet known structure of 3-methyl-6-(5-methyl-1,3,4-oxadiazol-2-yl)-[1,2,4]triazolo[4,3-a]pyridine (**2**) in 82% yield by a double [4+1] cyclocondensation reaction. The NMR data showed clearly that in this case an unexpected reactivity of the endocyclic nitrogen atom of the π -deficient pyridine ring towards the carbonyl carbon of the orthoester conducted to a geminated triazolopyridine nucleus.

Scheme 1. Synthetic route for oxadiazolyl-triazolo-pyridine (**2**). Reagents and conditions: (i) H₃CC(OEt)₃, reflux, 16 h.

In a second step and according to the literature,¹⁷⁻¹⁹ hydrazones can be obtained by refluxing ethanol or methanol as solvent, at 60 °C or at room temperature for 2 – 14 h depending on the reactivity of the carbonyl compound. Based on these literature data, we have performed several tests to obtain hydrazides (**4**), which varied the temperature, reaction time, solvent, and molar ratio. The most satisfactory results for the synthesis of hydrazides (**4a-g**) were obtained when we employed 6-hydrazinonicotinic acid hydrazide hydrate (**1**) and aryl/heteroaryl aldehydes (**3**) in a molar ratio 1:1 using ethanol as solvent, at 60 °C for 7 h (Scheme 2). These reaction conditions allowed us for the first time to differentiate the two nucleophilic centers at **1** and to isolate the compounds **4**, in good yields (64 – 94%) as stable solids by simple filtration under reduced pressure following by drying under vacuum in a desiccator containing P₂O₅. The structures of these new hydrazides (**4**) have then attached to the pyridine ring a stable hydrazone function and a free hydrazide function for further chemical modifications.

Scheme 2. Synthesis of hydrazinyl-nicotinohydrazide (**4a-g**). Reagents and conditions: (i) ArCHO (**3a-g**), EtOH, 60 °C, 7 h.

In an attempt to promote the synthesis of new heterocycles we have performed two types of cyclocondensation reactions, namely, [3+2] and [4+1] at the hydrazide function of compounds **4**. Thus, initially, a variety of 4-alkoxy-1,1,1-trifluoroalk-3-en-2-ones (**5**) was used to promote the synthesis of a new series of nine examples of 5-[(3-alkyl(aryl/heteroaryl)-5-trifluoromethyl-5-hydroxy-4,5-dihydro-1*H*-pyrazol-1-yl)carbonylpyridin-2-yl]aryl/hetero-arylaldehyde hydrazones (**6**) by the [3+2] cyclocondensation reactions. The reactions were based on the literature data¹⁵ and carried out in ethanol as solvent at reflux for 4 h. After the reaction time, the heterocycles **6** were isolated by simple filtration under reduced pressure in 45 – 83% yields (Scheme 3). The 4-methoxy-4-aryl/heteroaryl-1,1,1-trifluoroalk-3-en-2-ones (**5a-c**), which are readily available CCC synthetic blocks, were prepared from trifluoroacetylation reactions of enol ethers generated *in situ* from the respective acetophenone dimethyl acetals (for **5a-b**) or 2-acetilfuran dimethyl acetal (for **5c**) with trifluoroacetic anhydride, respectively, in the presence of pyridine as described in the literature.²⁰⁻²³

Finally, based on the first results from this work for the synthesis of compound **2**, we have employed the same reaction condition to obtain a 1,3,4-oxadiazole derivative from hydrazide (**4e**) by a [4+1] cyclocondensation reaction with triethylorthoacetate. This reaction condition allowed to obtain 2-(5-(5-methyl-1,3,4-oxadiazol-2-yl)-pyridin-2-yl)-4-chlorobenzaldehyde hydrazone (**7e**) in good yield (64%) when the reaction was carried out in triethylorthoacetate as solvent at reflux for 16 h.

The structures of 6-[2-(aryl/heteroarylidene)hydrazinyl] nicotinohydrazide (**4a-g**) were deduced from NMR experiments and compared with NMR data from other hydrazides formerly synthesized in the literature.¹⁵ Initially, compounds **4** show the ¹H NMR chemical shifts of the NH₂ at δ 4.42 ppm (broad singlet). The chemical shifts for the NH protons resonate at δ 9.61 and 11.24 ppm. These compounds show the CH=N proton at δ 8.19 ppm. Compounds **4a-g** present also the typical ¹³C chemical shifts of their carbons in average of δ 164.6 for (C=O), δ 136.8 (C=N), δ 158.1, 147.6, 140.4, 119.7, 105.0 ppm (pyridine ring).

Scheme 3. Synthetic route for pyrazolyl/oxazolyl-pyridine aryl/heteroaryl hydrazones (**6,7**). Reagents and conditions: (i) EtOH, reflux, 4 h; (ii) $\text{H}_3\text{CC}(\text{OEt})_3$, reflux, 16 h.

Compounds **6aa-6cc** show ^1H NMR chemical shifts of the diastereotopic methylene protons (H-4a and H-4b) as a characteristic of AB system and as a doublet at ca. δ 3.41 ppm and another doublet at ca. 3.76 ppm, with a *geminal* coupling constant in the range of $^2J = 18$ Hz. The absence of signals for NH and NH_2 protons from the hydrazide moiety of the precursors are also noted in the spectra of these compounds. The trifluoromethylated heterocycles present the typical ^{13}C NMR chemical shifts of pyrazoline ring carbons at δ 154.1 (C-3), 91.7 (C-5) with a coupling constant of $^2J = 33$ Hz, at 44.8 (C-4) and at 123.4 (CF_3) with $^1J = 286$ Hz. The ^{13}C chemical shifts of the pyridine ring carbons present signals in the range of 104.6, 120.3, 140.8, 149.2 and 158.5 ppm. The sp^2 -carbon of the hydrazone function for compounds **6** shows signals, such as singlet, in the average of 144.6 ppm.

Conclusions

The methodologies presented in this work allowed a chemical differentiation (chemoselectivity) of the *N*-reactivity in the precursor dinucleophilic functions (**1**). Furthermore, a facile and useful regioselective synthesis to obtain new hydrazides (**4**) was accomplished from the reaction of **1** with a broad scope of the selected aldehydes (**3**). We also demonstrated that hydrazides (**4**) can be used as the synthetic intermediates in [3+2] and [4+1] cyclocondensation reactions with two electrophilic molecules such as trifluoromethylated vinyl ketones **5** and triethylorthoacetate to isolate interesting new heterocyclic systems as pyrazolyl- and oxadiazolyl-pyridines (**6**) and (**7**), respectively.

Experimental Section

General. Unless otherwise indicated, all common reagents and solvents were used as obtained from commercial suppliers without further purification. All melting points were determined on a Reichert Thermovar apparatus. ¹H and ¹³C NMR spectra were acquired on a Bruker DPX 200 spectrometer (¹H at 200.13 MHz and ¹³C at 50.32 MHz) and Bruker DPX 400 (¹H at 400.13 MHz and ¹³C at 100.61 MHz) spectrometer, 5 mm sample tubes, 298 K, digital resolution ± 0.01 ppm, in DMSO-*d*₆, using TMS as internal reference (¹H and ¹³C). The CHN elemental analyses were performed on a Perkin Elmer 2400 CHN elemental analyzer (São Paulo University–USP/Brazil).

The ESI mass spectra were performed on an Agilent 6460 Triple Quadrupole connected to a 1200 series LC and equipped with a solvent degasser, binary pump, column oven, auto-sampler, and an ESI source. The Agilent QQQ 6460 tandem mass spectrometer was operated in the positive jet stream electrospray ionization (ESI) mode. Nitrogen was used as nebulizer, turbo (heater) gas, curtain gas, and collision activated dissociation gas. The capillary voltage was set at +3500 V and nozzle voltage was at +500 V. The ion source gas temperature was 300 °C with a flow rate of 5 L/min. The jet stream sheath gas temperature was 250 °C with a flow rate of 11 L/min. All samples were infused into the ESI source at a 5 μ L/min flow rate. Data were acquired in positive mode MS total ion scan mode (mass scan range *m/z* 50-650) and in positive MS/MS product ion scan mode. The mass spectra recorded were evaluated by the Qualitative Analysis software from Agilent Technologies.

General synthetic procedure for the synthesis of 3-methyl-6-(5-methyl-1,3,4-oxadiazol-2-yl)-[1,2,4]triazolo[4,3-*a*]pyridine (2**).** A mixture of 6-hydrazinonicotinic hydrazide hydrate **1** (1 mmol) and triethylorthoacetate (1.1 cm³, 6 mmol) was stirred for 16 h at reflux. The excess of orthoacetate was removed by evaporation under reduced pressure. The product **2** was obtained as solid and dried under vacuum in a desiccator containing P₂O₅.

3-Methyl-6-(5-methyl-1,3,4-oxadiazol-2-yl)-[1,2,4]triazolo[4,3-a]pyridine (2). White solid, yield 82%, mp 145 – 147 °C; ^1H NMR (400.13 MHz, DMSO- d_6): δ 8.93 (s, 1H, Py); 7.85 – 7.88 (m, 1H, Py); 7.76 – 7.79 (m, 1H, Py); 2.77 (s, 3H, CH₃); 2.61 (s, 3H, CH₃) ppm; ^{13}C NMR (100.61 MHz, DMSO- d_6): δ 163.9 (C_{oxa}–CH₃); 161.4 (C_{oxa}–Py); 148.2 (Py); 145.0 (C–CH₃); 144.9 (Py); 123.7 (Py); 115.9 (Py); 110.3 (Py); 10.3 (CH₃); 9.5 (CH₃) ppm; MS (ESI) m/z : [(M + H)⁺, 216.2]. Anal. Calcd for C₁₀H₉N₅O: C, 55.81; H, 4.22; N, 32.54. Found: C, 55.43; H, 4.21; N, 32.82.

General procedure for the synthesis of 6-[2-aryl/heteroarylmethylidenehydrazinyl]-nicotinohydrazide (4a-g).

A mixture of 6-hydrazinonicotinic hydrazide hydrate **1** (1 mmol) and aryl/heteroarylaldehydes **3a-g** (1 mmol) in 15 cm³ of ethanol was stirred for 7 h at 60 °C.

Products **4a-g** were obtained as solids, filtered under reduced pressure and dried under vacuum in a desiccator containing P₂O₅. Compounds **4a-g** presented a high degree of purity and did not undergo recrystallization.

6-(2-Benzylidenehydrazinyl)nicotinohydrazide (4a). White solid, yield 92%, mp 259 – 261 °C; ^1H NMR (400.13 MHz, DMSO- d_6): δ 11.27 (s, 1H, NH), 9.64 (s, 1H, NH), 8.59 – 8.60 (m, 1H, Py), 8.08 (s, 1H, H), 7.67 – 7.72 (m, 3H, Ph + Py), 7.37 – 7.45 (m, 3H, Ph), 7.24 – 7.29 (m, 1H, Py), 4.44 (s, 2H, NH₂) ppm; ^{13}C NMR (100.61 MHz, DMSO- d_6): δ 164.7 (CO), 158.3 (Py), 147.6 (Py), 140.5 (Py), 136.6 (C=N), 134.9 (Ph), 128.8 (Ph), 128.6 (Ph), 126.1 (Ph), 119.8 (Py), 105.2 (Py) ppm; MS (ESI) m/z : [(M + H)⁺, 256.1]. Anal. Calcd for C₁₃H₁₃N₅O: C, 61.17; H, 5.13; N, 27.43. Found: C, 61.56; H, 5.16; N, 27.72.

6-[2-(3-Nitrobenzylidene)hydrazinyl]nicotinohydrazide (4b). Yellow solid, yield 94%, mp 250 – 252 °C; ^1H NMR (400.13 MHz, DMSO- d_6): δ 11.54 (s, 1H, NH), 9.69 (s, 1H, NH), 8.62 – 8.63 (m, 1H, Py), 8.47 (s, 1H, H), 8.07 – 8.20 (m, 4H, Py + Ph), 7.66 – 7.74 (m, 1H, Ph), 7.30 – 7.35 (m, 1H, Py), 4.45 (s, 2H, NH₂) ppm; ^{13}C NMR (100.61 MHz, DMSO- d_6): δ 164.6 (CO), 158.0 (Py), 148.2 (Py), 147.6 (Ph), 138.0 (Py), 136.8 (Ph), 136.8 (C=N), 132.1 (Ph), 130.1 (Ph), 122.9 (Ph), 120.5 (Ph), 120.0 (Py), 105.5 (Py) ppm; MS (ESI) m/z : [(M + H)⁺, 301.1]. Anal. Calcd for C₁₃H₁₂N₆O₃: C, 52.00; H, 4.03; N, 27.99. Found: C, 52.35; H, 4.07; N, 27.72.

6-[2-(4-Methoxybenzylidene)hydrazinyl]nicotinohydrazide (4c). Yellow solid, yield 88%, mp 244 – 246 °C; ^1H NMR (400.13 MHz, DMSO- d_6): δ 10.92 (s, 1H, NH), 9.50 (s, 1H, NH), 8.57 (s, 1H, Py), 8.04 (s, 2H, Py + H), 7.61 – 7.63 (m, 2H, Ph), 7.19 – 7.22 (m, 1H, Py), 6.96 – 6.98 (m, 2H, Ph), 4.37 (s, 2H, NH₂), 3.79 (s, 3H, OCH₃) ppm; ^{13}C NMR (100.61 MHz, DMSO- d_6): δ 164.6 (CO), 159.8 (Ph), 158.2 (Py), 147.4 (Py), 140.5 (Py), 136.3 (C=N), 127.5 (Ph), 127.4 (Ph), 119.4 (Py), 114.0 (Ph), 104.0 (Py), 55.0 (OCH₃) ppm; MS (ESI) m/z : [(M + H)⁺, 286.2]. Anal. Calcd for C₁₄H₁₅N₅O₂: C, 58.94; H, 5.30; N, 24.55. Found: C, 58.88; H, 5.21; N, 24.43.

6-[2-(4-Methylbenzylidene)hydrazinyl]nicotinohydrazide (4d). White solid, yield 80%, mp 274 – 276 °C; ^1H NMR (400.13 MHz, DMSO- d_6): δ 11.14 (s, 1H, NH), 9.59 (s, 1H, NH), 8.57 (s, 1H, Py), 8.02 – 8.04 (m, 2H, Py + H), 7.57 – 7.59 (m, 2H, Ph), 7.21 – 7.24 (m, 3H, Ph + Py), 4.41 (s, 2H, NH₂), 2.32 (s, 3H, CH₃) ppm; ^{13}C NMR (100.61 MHz, DMSO- d_6): δ 164.6 (CO), 158.2 (Py), 147.4 (Py), 140.6 (Py), 138.3 (C=N), 136.4 (Ph), 132.1 (Ph), 129.0 (Ph), 125.9 (Ph),

119.6 (Py), 105.0 (Py), 20.6 (CH₃) ppm; MS (ESI) m/z : [(M + H)⁺, 270.1]. Anal. Calcd for C₁₄H₁₅N₅O: C, 62.44; H, 5.61; N, 26.01. Found: C, 62.82; H, 5.53; N, 25.73.

6-[2-(4-Chlorobenzylidene)hydrazinyl]nicotinohydrazide (4e). White solid, yield 90%, mp 280 – 282 °C; ¹H NMR (400.13 MHz, DMSO-*d*₆): δ 11.30 (s, 1H, NH), 9.62 (s, 1H, NH), 8.59 (s, 1H, Py), 8.04 – 8.06 (s, 2H, Py + H), 7.72 (d, *J* = 8, 2H, Ph), 7.47 (d, *J* = 8, 2H, Ph), 7.26 – 7.28 (m, 1H, Py), 4.37 (s, 2H, NH₂) ppm; ¹³C NMR (100.61 MHz, DMSO-*d*₆): δ 164.6 (CO), 158.1 (Py), 147.5 (Py), 139.2 (Py), 136.5 (C=N), 133.7 (Ph), 133.0 (Ph), 128.5 (Ph), 127.5 (Ph), 119.7 (Py), 105.2 (Py) ppm; MS (ESI) m/z : [(M + H)⁺, 290.0]. Anal. Calcd for C₁₃H₁₂ClN₅O: C, 53.89; H, 4.17; N, 24.17. Found: C, 54.27; H, 4.15; N, 24.31.

6-[2-(Naphth-1-ylmethylidene)hydrazinyl]nicotinohydrazide (4f). Yellow solid, yield 85%, mp 250 – 252 °C; ¹H NMR (400.13 MHz, DMSO-*d*₆): δ 11.36 (s, 1H, NH), 9.68 (s, 1H, NH), 8.82 (s, 1H, Py), 8.65 (s, 1H, H), 8.14 – 8.15 (m, 1H, Py), 7.95 – 8.09 (m, 4H, Naphthyl), 7.55 – 7.65 (m, 3H, Naphthyl), 7.30 – 7.34 (m, 1H, Py), 4.48 (s, 2H, NH₂) ppm; ¹³C NMR (100.61 MHz, DMSO-*d*₆): δ 164.8 (CO), 158.4 (Py), 147.9 (Py), 140.3 (Py), 137.0 (C=N), 133.6 (Naphthyl), 130.1 (Naphthyl), 129.9 (Naphthyl), 129.4 (Naphthyl), 128.8 (Naphthyl), 127.1 (Naphthyl), 126.1 (Naphthyl), 125.9 (Naphthyl), 125.7 (Naphthyl), 123.6 (Naphthyl), 119.6 (Py), 105.4 (Py) ppm; MS (ESI) m/z : [(M + H)⁺, 306.1]. Anal. Calcd for C₁₇H₁₅N₅O: C, 66.87; H, 4.95; N, 22.94. Found: C, 66.59; H, 4.93; N, 23.14.

6-[2-(Furan-2-ylmethylidene)hydrazinyl]nicotinohydrazide (4g). Brown solid, yield 64%, mp 226 – 228 °C; ¹H NMR (400.13 MHz, DMSO-*d*₆): δ 11.20 (s, 1H, NH), 9.57 (s, 1H, NH), 8.60 (s, 1H, Py), 8.01 – 8.07 (m, 2H, Py + H), 7.74 (s, 1H, Furyl), 7.15 – 7.17 (m, 1H, Furyl), 6.76 – 6.77 (m, 1H, Py), 6.58 – 6.59 (m, 1H, Furyl), 4.42 (s, 2H, NH₂) ppm; ¹³C NMR (100.61 MHz, DMSO-*d*₆): δ 164.6 (CO), 158.0 (Py), 149.9 (Furyl), 147.4 (Py), 143.7 (Py), 136.5 (C=N), 130.9 (Furyl), 119.8 (Py), 111.7 (Furyl), 110.6 (Furyl), 105.0 (Py) ppm; MS (ESI) m/z : [(M + H)⁺, 246.0]. Anal. Calcd for C₁₁H₁₁N₅O₂: C, 53.87; H, 4.52; N, 28.56. Found: C, 54.22; H, 4.70; N, 28.84.

General synthetic procedure for the synthesis of 5-[(3-alkyl(aryl/heteroaryl)-5-trifluoromethyl-5-hydroxy-4,5-dihydro-1H-pyrazol-1-yl)carbonylpyridin-2-yl]aryl-

heteroarylaldehyde hydrazones (6aa – 6ac, 6ba – 6bc, 6ca – 6cc). A mixture of 1,1,1-trifluoro-4-alkyl(aryl/heteroaryl)-4-methoxy-3-alken-2-one (**5a-c**) (1 mmol) and 6-[2-aryl/heteroarylmethylidenehydrazino]nicotinic hydrazide (**4a-c**) (1 mmol) in 15 cm³ of ethanol was refluxed under stirring for 4 h. Products **6aa-6cc** were obtained as solids, filtered under reduced pressure, and dried in a desiccator containing P₂O₅ under vacuum. All compounds presented a high degree of purity and did not undergo recrystallization.

5-[(5-Trifluoromethyl-5-hydroxy-3-methyl-4,5-dihydro-1H-pyrazol-1-yl)- carbonylpyridin-2-yl]benzaldehyde hydrazine (6aa). White solid, yield 79%, mp 102 – 104 °C; ¹H NMR (400.13 MHz, DMSO-*d*₆): δ 11.34 (s, 1H, NH), 8.57 (s, 1H, Py), 8.11 (s, 1H, H), 8.01 – 8.06 (m, 1H, Py), 7.68 – 7.79 (m, 3H, Ph), 7.39 – 7.43 (m, 2H, Ph), 7.25 – 7.30 (m, 1H, Py), 3.49 (d, 1H, *J*_{H4-H4'} 19 Hz, 1H, H4'-Pz), 3.09 (d, 1H, *J*_{H4-H4'} 19 Hz, 1H, H4'-Pz), 2.02 (s, 3H, CH₃) ppm; ¹³C

NMR (100.61 MHz, DMSO-*d*₆): δ 165.1 (CO), 157.9 (Py), 154.2 (C3-Pz), 150.2 (Py), 149.9 (C=N), 143.9 (Py), 139.9 (Ph), 131.5 (Ph), 123.2 (q, 1J 286 Hz, CF₃), 120.9 (Py), 111.8 (Ph), 110.9 (Ph), 104.5 (Py), 91.5 (q, 1J 33 Hz, C5 – Pz), 47.2 (CH₂ – Pz), 15.0 (CH₃) ppm; MS (ESI) *m/z*: [(M + H)⁺, 392.1]. Anal. Calcd for C₁₈H₁₆F₃N₅O₂: C, 55.24; H, 4.12; N, 17.90. Found: C, 55.46; H, 4.43; N, 17.58.

5-[(5-Trifluoromethyl-5-hydroxy-3-phenyl-4,5-dihydro-1H-pyrazol-1-yl)- carbonylpyridin-2-yl]benzaldehyde hydrazone (6ab). White solid, yield 58%, mp 219 – 221 °C; ^1H NMR (400.13 MHz, DMSO-*d*₆): δ 11.38 (s, 1H, NH), 8.69 (s, 1H, Py), 8.09 – 8.21 (m, 2H, Py + H), 7.72 – 7.79 (m, 5H, Ph), 7.33 – 7.51 (m, 6H, Ph + Py), 3.96 (d, 1H, $J_{\text{H4-H4}'} = 19$ Hz, 1H, H4-Pz), 3.62 (d, 1H, $J_{\text{H4-H4}'} = 19$ Hz, 1H, H4'-Pz) ppm; ^{13}C NMR (100.61 MHz, DMSO-*d*₆): δ 164.9 (CO), 158.1 (Py), 151.7 (C3-Pz), 150.4 (Py), 141.2 (C=N), 139.3 (Py), 134.7 (Ph), 130.3 (Ph), 129.9 (Ph), 128.7 (Ph), 128.5 (Ph), 128.4 (Ph), 126.3 (Ph), 126.0 (Ph), 123.1 (q, 1J 286 Hz, CF₃), 120.7 (Py), 104.6 (Py), 92.1 (q, 1J 33 Hz, C5 – Pz), 43.6 (CH₂ – Pz) ppm; MS (ESI) *m/z*: [(M + H)⁺, 454.2]. Anal. Calcd for C₂₃H₁₈F₃N₅O₂: C, 60.93; H, 4.00; N, 15.45. Found: C, 61.31; H, 4.21; N, 15.71.

5-[(5-Trifluoromethyl-5-hydroxy-3-(2-furyl)-4,5-dihydro-1H-pyrazol-1-yl)- carbonylpyridin-2-yl]benzaldehyde hydrazone (6ac). Yellow solid, yield 45%; mp 212 – 214 °C; ^1H NMR (400.13 MHz, DMSO-*d*₆): δ 11.18 (s, 1H, NH), 8.66 (s, 1H, Py), 8.10 – 8.14 (m, 2H, Py + H), 7.86 (s, 1H, Furyl), 7.70 – 7.71 (m, 2H, Ph), 7.33 – 7.44 (m, 3H, Ph), 7.29 – 7.31 (m, 1H, Furyl), 7.06 (s, 1H, Furyl), 6.65 – 6.67 (m, 1H, Py), 3.82 (d, 1H, $J_{\text{H4-H4}'} = 18$ Hz, H4-Pz), 3.49 (d, 1H, $J_{\text{H4-H4}'} = 18$ Hz, 1H, H4'-Pz) ppm. ^{13}C NMR (100.61 MHz, DMSO-*d*₆): δ 164.8 (CO), 158.2 (Py), 150.5 (C3-Pz), 145.7 (Py), 145.1 (C=N), 143.1 (Py), 141.2 (Furyl), 139.5 (Furyl), 134.8 (Ph), 128.9 (Ph), 128.6 (Ph), 126.2 (Ph), 125.9 (q, 1J 286 Hz, CF₃), 120.7 (Py), 114.6 (Furyl), 112.1 (Furyl), 104.7 (Py), 91.7 (q, 1J 33 Hz, C5-Pz), 43.5 (CH₂-Pz) ppm. MS (ESI) *m/z*: [(M + H)⁺, 444.1]. Anal. Calcd for C₂₁H₁₆F₃N₅O₃: C, 56.89; H, 3.64; N, 15.80. Found: C, 57.07; H, 3.88; N, 16.03.

5-[(5-Trifluoromethyl-5-hydroxy-3-methyl-4,5-dihydro-1H-pyrazol-1-yl)- carbonylpyridin-2-yl]3-nitrobenzaldehyde hydrazone (6ba). Yellow solid, yield 73%, mp 246 – 248 °C; ^1H NMR (400.13 MHz, DMSO-*d*₆): δ 11.59 (s, 1H, NH), 8.59 – 8.60 (m, 1H, Py), 8.49 (s, 1H, H), 8.07 – 8.22 (m, 2H, Ph), 8.07 (s, 1H, Py), 7.80 (s, 1H, Ph), 7.69 – 7.73 (m, 1H, Ph), 7.33 – 7.35 (m, 1H, Py), 3.49 (d, 1H, $J_{\text{H4-H4}'} = 19$ Hz, H4-Pz), 3.11 (d, 1H, $J_{\text{H4-H4}'} = 19$ Hz, 1H, H4'-Pz), 2.03 (s, 3H, CH₃) ppm; ^{13}C NMR (100.61 MHz, DMSO-*d*₆): δ 164.9 (CO), 157.7 (Py), 154.1 (C3-Pz), 150.1 (Py), 148.1 (C=N), 139.3 (Py), 138.4 (Ph), 136.7 (Ph), 132.0 (Ph), 129.9 (Ph), 123.2 (q, 1J 286 Hz, CF₃), 122.7 (Ph), 121.5 (Ph), 119.9 (Py), 104.9 (Py), 91.4 (q, 1J 33 Hz, C5-Pz), 47.1 (CH₂-Pz), 15.0 (CH₃) ppm; MS (ESI) *m/z*: [(M + H)⁺, 437.0]. Anal. Calcd for C₁₈H₁₅F₃N₆O₄: C, 49.55; H, 3.46; N, 19.26. Found: C, 49.90; H, 3.21; N, 19.42.

5-[(5-Trifluoromethyl-5-hydroxy-3-phenyl-4,5-dihydro-1H-pyrazol-1-yl)-carbonylpyridin-2-yl]3-nitrobenzaldehyde hydrazone (6bb). Yellow solid, yield 76%, mp 263 – 265 °C; ^1H NMR (400.13 MHz, DMSO-*d*₆): δ 11.67 (s, 1H, NH), 8.69 – 8.70 (m, 1H, Py), 8.51 (s, 1H, H), 8.15 – 8.24 (m, 5H, Ph), 7.76 – 7.80 (m, 2H, Ph), 7.76 – 7.80 (m, 2H, Ph), 7.68 – 7.72 (m, 1H,

Py), 7.49 – 7.52 (m, 2H, Ph), 7.39 – 7.43 (m, 1H, Py), 3.97 (d, 1H, $J_{\text{H4-H4'}}$ 19 Hz, 1H, H4-Pz), 3.63 (d, 1H, $J_{\text{H4-H4'}}$ 19 Hz, H4'-Pz) ppm; ^{13}C NMR (100.61 MHz, DMSO- d_6): δ 164.7 (CO), 157.7 (Py), 151.7 (C3-Pz), 150.2 (Py), 148.1 (C=N), 139.4 (Py), 138.5 (Ph), 136.6 (Ph), 132.0 (Ph), 130.3 (Ph), 129.9 (Ph), 128.5 (Ph), 126.3 (Ph), 123.6 (Ph), 123.0 (q, 1J 285 Hz, CF_3), 122.7 (Ph), 121.4 (Ph), 119.9 (Py), 104.9 (Py), 92.1 (q, 1J 33 Hz, C5-Pz), 43.6 (CH_2 -Pz) ppm; MS (ESI) m/z : $[(\text{M} + \text{H})^+]$, 499.0]. Anal. Calcd for $\text{C}_{23}\text{H}_{17}\text{F}_3\text{N}_6\text{O}_4$: C, 52.68; H, 3.45; N, 17.98. Found: C, 52.34; H, 3.44; N, 17.65.

5-[(5-Trifluoromethyl-5-hydroxy-3-(2-furyl)-4,5-dihydro-1H-pyrazol-1-yl)-

carbonylpyridin-2-yl]3-nitrobenzaldehyde hydrazone (6bc). Yellow solid, yield 83%, mp 252 – 254 °C; ^1H NMR (400.13 MHz, DMSO- d_6): δ 11.48 (s, 1H, NH), 8.67 – 8.68 (m, 1H, Py), 8.48 (s, 1H, H), 8.24 (s, 1H, Py), 8.13 – 8.16 (m, 3H, Ph), 7.87 (s, 1H, Furyl), 7.68 – 7.72 (m, 1H, Ph), 7.35 – 7.37 (m, 1H, Furyl), 7.07 – 7.08 (m, 1H, Py), 6.66 – 6.67 (m, 1H, Furyl), 3.83 (d, 1H, $J_{\text{H4-H4'}}$ 18 Hz, H4-Pz), 3.50 (d, 1H, $J_{\text{H4-H4'}}$ 18 Hz, H4'-Pz) ppm; ^{13}C NMR (100.61 MHz, DMSO- d_6): δ 164.6 (CO), 157.8 (Py), 150.2 (C3-Pz), 148.1 (Py), 145.5 (C=N), 145.0 (Py), 143.1 (Furyl), 139.4 (Furyl), 138.6 (Ph), 136.6 (Ph), 132.0 (Ph), 129.9 (Ph), 122.7 (Ph), 122.9 (q, 1J 286 Hz, CF_3), 121.6 (Ph), 119.9 (Py), 114.2 (Furyl), 111.9 (Furyl), 104.9 (Py), 91.6 (q, 1J 33 Hz, C5-Pz), 43.4 (CH_2 -Pz) ppm; MS (ESI) m/z : $[(\text{M} + \text{H})^+]$, 489.1]. Anal. Calcd for $\text{C}_{21}\text{H}_{15}\text{F}_3\text{N}_6\text{O}_5$: C, 52.33; H, 3.33; N, 17.72. Found: C, 52.37; H, 3.16; N, 17.63.

5-[(5-Trifluoromethyl-5-hydroxy-3-methyl-4,5-dihydro-1H-pyrazol-1-yl)-carbonylpyridin-

2-yl]4-methoxybenzaldehyde hydrazone (6ca). White solid, yield 77%, mp 194 – 196 °C; ^1H NMR (400.13 MHz, DMSO- d_6): δ 11.10 (s, 1H, NH), 8.56 (s, 1H, Py), 7.99 – 8.06 (m, 2H, Py + H), 7.63 (d, J 8, 2H, Ph), 7.19 – 7.24 (m, 1H, Py), 6.98 (d, J 8, 2H, Ph), 3.8 (s, 3H, OCH_3), 3.47 (d, 1H, $J_{\text{H4-H4'}}$ 19 Hz, H4-Pz), 3.10 (d, 1H, $J_{\text{H4-H4'}}$ 19 Hz, 1H, H4'-Pz), 2.02 (s, 3H, CH_3) ppm; ^{13}C NMR (100.61 MHz, DMSO- d_6): δ 165.0 (CO), 160.0 (Py), 158.2 (C3-Pz), 154.2 (Ph), 150.4 (Py), 141.1 (C=N), 139.3 (Py), 127.7 (Ph), 123.2 (q, 1J = 286 Hz, CF_3), 120.4 (Py), 114.1 (Ph), 104.4 (Py), 91.4 (q, 1J = 33 Hz, C5-Pz), 55.1 (OCH_3), 47.2 (CH_2 -Pz), 15.1 (CH_3) ppm; MS (ESI) m/z : $[(\text{M} + \text{H})^+]$, 422.1]. Anal. Calcd for $\text{C}_{18}\text{H}_{18}\text{F}_3\text{N}_5\text{O}_3$: C, 54.16; H, 4.31; N, 16.62. Found: C, 54.35; H, 4.29; N, 16.58.

5-[(5-Trifluoromethyl-5-hydroxy-3-phenyl-4,5-dihydro-1H-pyrazol-1-yl)-carbonylpyridin-

2-yl]4-methoxybenzaldehyde hydrazone (6cb). White solid, yield 79%, mp 217 – 219 °C; ^1H NMR (400.13 MHz, DMSO- d_6): δ 11.09 (s, 1H, NH), 8.68 (s, 1H, Py), 8.14 – 8.17 (m, 1H, Py), 8.10 (s, 1H, H), 7.75 – 7.77 (m, 2H, Ph), 7.66 (d, J 8, 2H, Ph), 7.48 – 7.49 (m, 3H, Ph), 7.28 – 7.30 (m, 1H, Py), 6.99 (d, J 8, 2H, Ph), 3.92 (d, 1H, $J_{\text{H4-H4'}}$ 19 Hz, H4-Pz), 3.80 (s, 3H, OCH_3), 3.61 (d, 1H, $J_{\text{H4-H4'}}$ 19 Hz, H4'-Pz) ppm; ^{13}C NMR (100.61 MHz, DMSO- d_6): δ 165.0 (CO), 160.0 (Py), 158.2 (C3-Pz), 151.7 (Ph), 150.6 (Py), 141.3 (C=N), 139.4 (Py), 130.4 (Ph), 130.0 (Ph), 128.6 (Ph), 127.7 (Ph), 127.4 (Ph), 126.4 (Ph), 123.2 (q, 1J = 286 Hz, CF_3), 120.3 (Py), 114.1 (Ph), 104.5 (Py), 92.2 (q, 1J = 33 Hz, C5-Pz), 55.0 (OCH_3), 43.7 (CH_2 -Pz) ppm; MS (ESI) m/z : $[(\text{M} + \text{H})^+]$, 484.1]. Anal. Calcd for $\text{C}_{24}\text{H}_{20}\text{F}_3\text{N}_5\text{O}_3$: C, 59.63; H, 4.17; N, 14.49. Found: C, 59.94; H, 4.21; N, 14.64.

5-[(5-Trifluoromethyl-5-hydroxy-3-(2-furyl)-4,5-dihydro-1H-pyrazol-1-yl)-

carbonylpyridin-2-yl]4-methoxybenzaldehyde hydrazone (6cc). White solid, yield 76%, mp 226 – 228 °C; ¹H NMR (400.13 MHz, DMSO-*d*₆): δ 11.10 (s, 1H, NH), 8.64 (s, 1H, Py), 8.07 – 8.10 (m, 2H, Py + H), 7.72 – 7.73 (m, 1H, Furyl), 7.65 (d, *J* 8, 2H, Ph), 7.55 – 7.56 (m, 1H, Furyl), 7.24 – 7.28 (m, 1H, Furyl), 7.16 – 7.18 (m, 1H, Py), 6.99 (d, *J* 8, 2H, Ph), 3.93 (d, 1H, *J*_{H4-H4'} = 18 Hz, H4-Pz), 3.80 (s, 3H, OCH₃), 3.59 (d, 1H, *J*_{H4-H4'} 19 Hz, H4'-Pz) ppm; ¹³C NMR (100.61 MHz, DMSO-*d*₆): δ 164.4 (CO), 159.9 (Py), 158.1 (C3-Pz), 150.5 (Ph), 147.5 (Py), 141.2 (C=N), 139.3 (Py), 133.0 (Furyl), 130.6 (Furyl), 129.7 (Furyl), 128.3 (Furyl), 127.9 (Ph), 127.6 (Ph), 123.0 (q, ¹*J* 286 Hz, CF₃), 120.1 (Py), 114.0 (Ph), 104.3 (Py), 92.1 (q, ¹*J* 33 Hz, C5-Pz), 55.4 (OCH₃), 44.2 (CH₂-Pz) ppm; MS (ESI) *m/z*: [(M + H)⁺, 474.2]. Anal. Calcd for C₂₂H₁₈F₃N₅O₄: C, 55.82; H, 3.83; N, 14.79. Found: C, 56.17; H, 4.21; N, 14.46.

General synthetic procedure for the synthesis of 2-(5-(5-methyl-1,3,4-oxadiazol-2-yl)-pyridin-2-yl)-4-chlorobenzaldehyde hydrazone (7e). A mixture of 6-[2-aryl/heteroarylmethylidenehydrazino]nicotinic hydrazide (**4e**) (1 mmol) and triethylorthoacetate (1.1 cm³, 6 mmol) was stirred for 16 h at reflux. The excess of orthoacetate was removed by evaporation under reduced pressure. The product **7e** was obtained as solid and dried under vacuum in a desiccator containing P₂O₅.

2-(5-(5-Methyl-1,3,4-oxadiazol-2-yl)-pyridin-2-yl)-4-chlorobenzaldehyde hydrazone (7e)

Brown solid, yield 64%, mp 161 – 163 °C; ¹H NMR (400.13 MHz, DMSO-*d*₆): δ 11.35 (s, 1H, NH), 8.68 (s, 1H, Py), 8.12 – 8.14 (m, 2H Py + H), 7.70 – 7.73 (m, 2H, Ph), 7.44 – 7.47 (m, 2H, Ph), 7.37 – 7.40 (m, 1H, Py), 2.55 (s, 3H, CH₃) ppm; ¹³C NMR (100.61 MHz, DMSO-*d*₆): δ 162.8 (C₅-Oxz), 162.5 (C₂-Oxz), 158.2 (Py); 146.4 (Py), 139.9 (Py), 135.5 (C=N), 128.6 (Ph), 128.5 (Ph), 127.7 (Ph), 127.5 (Ph), 119.8 (Py), 106.3 (Py), 10.3 (CH₃-Oxz) ppm; MS (ESI) *m/z*: [(M + H)⁺, 314.1]. Anal. Calcd for C₁₅H₁₂ClN₅O: C, 57.42; H, 3.86; N, 22.32. Found: C, 57.39; H, 4.01; N, 22.65.

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